



Internal standard = 23.8 mg

Sample = 27.6 mg

Molar ratio = $(1.22/1)/(1/1) = 1.22$

Molar ratio = $(1.14/1)/(1/1) = 1.14$

MW of **6** = 198.27

MW of internal standard = 226.14

% wt = 92.2%

% wt = 86.2%

Averaged: % wt = 89.2%

$$\% \text{ wt} = \left(\frac{\text{mass of std}}{\text{MW of std}} \times \text{molar ratio} \times \text{MW of cpd} \right) \div \text{mass of cpd}$$